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$11\text{H} + X \rightarrow 63\text{Li} + 42\text{He}$  A)  $94\text{Li}$  B)  $94\text{Be}$

C)  $105\text{Be}$  D)  $106\text{C}$  The particle

represented by X is A)  $53\text{Fe}$  B)  $137\text{Cs}$

C)  $198\text{Au}$  D)  $220\text{Fr}$  2. Which isotope will

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Q. Barium-122 has a half-life of 2 minutes. A fresh sample weighing 80 g was obtained. If it takes 10 minutes to set up an experiment using barium-122, how much barium-122 will be left when the experiment begins?

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