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World Of Chemistry Chapter 16 Test

Chapter 16 Acids and Bases. World of Chemistry 168 Copyright Houghton Mifflin Company. All rights reserved. Chapter 16 Acids and Bases. 1. Acids were recognized primarily from their sour taste. Bases were recognized from their bitter taste and slippery feel on skin. 2. In the Arrhenius definition, an acid is a substance that produces hydrogen ions (H^+) when dissolved in water, whereas a base is a substance that produces hydroxide ions (OH^-) in aqueous solution.

Chapter 16 Acids and Bases

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Episode 16 - The Proton in Chemistry. The World of Chemistry. Episode 16 - The Proton in Chemistry. 1. Lime (calcium hydroxide) is the fifth most produced chemical in the United States. What are some uses for lime?

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Using the techniques explained in Chapter 7, we can balance the equation. $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2(g) + 4H_2O(g)$ Check: 3 C, 8 H, 10 O \rightarrow 3 C, 8 H, 10 O This equation can be interpreted in terms of molecules as follows: 1 molecule of C_3H_8 ! 5 molecules of O_2 \rightarrow 3 molecules of CO_2 ! 4 molecules of H_2O

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12. 1 mol O = 16.00 g O = 6.02×10^{23} O atoms 1 O atom 16.00 g
O 6.022×10^{23} O atoms = 2.66 $\times 10^{-23}$ g O 13. 0.50 mol O atoms
16.00 g O 1 mol = 8.0 g O 4 mol H atoms 1.008 g 1 mol = 4 g H
Half a mole of O atoms weighs more than 4 moles of H atoms.

Chapter 6 Chemical Composition

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